BIOLOGICALLY ACTIVE COMPOUNDS FROM Limonium

Gmelinii AND L. Popovii. I

UDC 547.972

L. M. Korul'kina,¹ E. E. Shul'ts,² G. E. Zhusupova,¹ Zh. A. Abilov,¹ K. B. Erzhanov,¹ and M. I. Chaudri³

Compositions of ethylacetate fractions of roots and aerial parts of Limonium Gmelinii and L. Popovii (*Plumbaginaceae*) were studied. 3,5,7,3',4',6'-Hexahydroxyflavone and myricetin 3-O- α -L-(2'-galloyl)arabopyranoside, the structures of which were established using chemical transformations and spectral data, were isolated for the first time.

Key words: *Limonium Gmelinii*, *L. Popovii*, Plumbaginaceae, flavonoids, 3,5,7,3',4',6'-hexahydroxyflavone, myricetin 3-O- α -L-(2'-galloyl)arabinopyranoside

Concentrated methanolic extracts of roots and aerial parts of *Limonium Gmelinii* and *L. Popovii* were successively fractionated with petroleum ether, chloroform, diethylether, ethylacetate, and butanol.

The ethylacetate fractions were chromatographed over polyamide columns with elution by CHCl₃ and CHCl₃:CH₃OH mixtures (9:1, 1:1, and 1:9). Identical fractions were combined and rechromatographed over polyamide, silica gel, and Sephadex LH-20.

Roots of Gmelin sea-lavender afforded D-(+)-galactose (1) [1]; gallic (2) [2], syringic (3) [3], and ellagic (4) [4] acids; quercetin (5) [5, 6]; rutin (6) [7]; myricetin (7) [8] and its glycosides: myricetrin (8) [9, 10], 7-O- α -L-rhamnopyranoside (9) [11], rutinoside (10) [7], 3-O- α -L-arabopyranoside (11) [12], 3-O- β -D-xyloside (12) [13], 3-O- α -L-(2"-O- α rhamnopyranosyl)rhamnopyranoside (13) [14], and (-)-epigallocatechin-3-O-gallate (14) [15] in addition to the new, previously undescribed 3,5,7,3',4',6'-hexahydroxyflavone (15) and myricetin 3-O- α -L-(2'-galloyl)arabopyranoside (16).

We identified 2, 5, 7, 8, 14, 15, 16, and myricetin 3-O- β -D-(6"-galloyl)galactopyranoside (17) [12] in the aerial part of Gmelin sea-lavender.

Roots of Popov sea-lavender afforded 2, 5, and 7; the aerial part, 2, 5, 7, 8, 17, and myricetin 3-O- β -D-galactopyranoside (18) [9, 16], 3-O- β -D-glucopyranoside (19) [17], and 3-O- β -D-(6"-galloyl)glucopyranoside (20) [18].

All compounds were identified by IR, mass, PMR, and ¹³C NMR spectra; chemical transformations, and comparison with the literature.

3,5,7,3',4',6'-Hexahydroxyflavone is a new and previously undescribed compound, yellowish-green, mp 306-308°C. Qualitative reactions and chromatographic behavior [paper (PC) and thin-layer (TLC)] indicated that it is a flavone with a free 3-OH group [19].

Two doublets in the PMR with spin—spin coupling constants (SSCC) 2 Hz near 6.21 and 6.45 ppm are consistent with *meta*-substitution of ring A [21]. A doublet with SSCC 1.8 Hz near 7.69 ppm and a broad singlet near 7.8 ppm may belong to H-2' and H-5' or H-3' and H-6' of ring B.

¹⁾ Al-Farabi Kazakh National University, 480012, Almaty, ul. Karasai Batyra, 95 a, fax (3272) 74 26 09, e-mail: lira90@list.ru; 2) N. N. Vorozhtsov Novosibirsk Institute of Organic Chemistry, Siberian Division, Russian Academy of Sciences, 630090, Novosibirsk, pr. Lavrent'eva, 9, fax (3832) 34 47 52; 3) University Scientific Institute of Chemistry, Karachi (Pakistan). Translated from Khimiya Prirodnykh Soedinenii, No. 5, pp. 383-387, September-October, 2004. Original article submitted June 7, 2004.



Scheme 1. Proposed fragmentation scheme of 3,5,7,3',4',6'-hexahydroxyflavone.

The ¹³C NMR spectrum contains signals for 15 C atoms. The positions of signals for ring A agree completely with the literature for a 5,7-dihydroxysubstituted flavonol with a free 3-OH. C-6 is near 98.25 ppm; C-8, 93.347 ppm [22]. A doublet for C-2' appears near 111.52 ppm; C-5', 114.09 also as a doublet. *Ortho*-substituted C-3' and C-4' resonate near 141.18 and 140.11 ppm; C-6', 146.13 ppm.

Bathochromic shifts of bands induced by $AlCl_3$ in the UV spectrum are retained upon addition of HCl and indicate free C-3 and C-5 hydroxyls. Bathochromic shifts of the second band with NaOAc and of the first with H_3BO_3 are indicative of a free 7-OH in ring A and an *ortho*-dihydroxy in ring B.

The molecular weight (EI method) of 318 corresponds with the empirical formula $C_{15}H_{10}O_8$. A characteristic fragment with m/z 152 arises from retro-decomposition according to the proposed fragmentation scheme (Scheme 1) [20].

Myricetin 3-O- α -L-(2'-galloyl)arabinopyranoside, [M]⁺ 603, C₂₇H₂₃O₁₆, is also new and not previously described. It is a yellow compound, mp 182-184°C. Qualitative reactions and chromatographic behavior (PC and TLC) indicated that it is a flavonol glycoside with a substituted 3-OH [19].

Stepwise hydrolysis with time and PC after 20 min detected gallic acid and an intermediate glycoside; after 60 min, the aglycon, gallic acid, and arabinose. The chromatographic behavior, PMR spectrum, and lack of melting-point depression in a mixed sample identified the aglycon as myricetin. PC using systems 2 and 4 detected gallic acid and arabinose in the hydrolysis products.

Absorption bands in the IR spectrum near 3300-3400 cm⁻¹ (OH), 1654-1670 (C=O), and three strong bands at 1028, 1070, and 1086 indicate the pyranose form of the sugar. A band near 836 cm⁻¹ corresponds with the α -glycoside bond.

The mass spectrum (positive-ion FAB) gave a peak for the molecular ion with m/z 604, corresponding with $C_{27}H_{24}O_{16}$, and a peak for an ion with m/z 318 (EI), $C_{15}H_{10}O_8$. The presence of gallic acid in the structure was confirmed by formation of a fragment with m/z 450, corresponding to loss of acid, and a fragment with m/z 318, corresponding to loss of arabinose. A characteristic fragment with m/z 152 corresponds to retro-decomposition of flavonoids [20]. The proposed fragmentation scheme of myricetin 3-O- α -L-(2'-galloyl)arabinopyranoside is given below.

A 2H-singlet near 7.08 ppm corresponds to gallic-acid protons; a 1H-doublet near 5.17 with SSCC 6.6 Hz, to H' of α -L-arabinose. The attachment site of gallic acid and the sugar was determined as C-3 using ¹³C NMR spectra and HMBC twodimensional spectra.

The ¹³C NMR spectrum exhibits 27 C atoms. The positions of signals for rings A and B correspond with literature data for myricetin, arabinose, and gallic acid [22].



Scheme 2. Proposed fragmentation scheme of myricetin 3-O- α -L-(2'-galloyl)arabinopyranoside.

EXPERIMENTAL

We used PC and TLC and solvent systems: $CHCl_3:CH_3OH(95:5, 90:10, 85:15)(1)$, *n*-butanol: $CH_3CO_2H:H_2O(40:12, 5:29)(2)$, $CH_3CO_2H(2\%)(3)$, and *n*-butanol: $C_5H_5N:H_2O(6:4:3)(4)$.

TLC was performed on Silicagel 60 F 254 (Merck) plates with identification in UV light and reaction with cerium sulfate at 100-105°C in system 1; PC, on Filtrak No. 3 chromatography paper with identification in UV light and reaction with ammonia vapor, $AlCl_3$, and ferric-ammonium alum in systems 2 and 3. Carbohydrates were detected using *o*-toluidine with heating for 3-5 min at 90-100°C in system 4.

Extraction and Isolation of Phenolic Acids and Flavonoids. Air-dried raw material was ground to 1-3 mm and extracted three times with CH_3OH at room temperature (1:4 material:solvent ratio). The methanol extract was evaporated to a small volume and fractionated successively with petroleum ether, $CHCl_3$, diethylether, ethylacetate, and butanol.

The ethylacetate extracts were chromatographed over polyamide columns, silica gel (Silicagel 60, 70-230 mesh, Merck), and Sephadex (LH-20) with elution by $CHCl_3$, $CHCl_3:CH_3OH$ mixtures (19:1, 9:1, 4:1, and 1:1), and CH_3OH . Elution from Sephadex columns used $H_2O:CH_3OH$ mixtures (95:5, 9:1, 5:1, and 1:1) and CH_3OH . Pure compounds were isolated by rechromatography using the identical columns and eluents.

Acid hydrolysis was performed in HCl (2 N) for 2 h on a boiling-water bath. The aglycon was extracted by ethylacetate. Sugar was identified by PC using system 3.

Stepwise hydrolysis with time was performed in HCl (0.2 N) in ethanol for 2 h on a boiling-water bath. The hydrolysis products were identified by PC using system 2.

IR spectra were recorded on JASCO IR A-1 and Shimadzu IR-460 instruments; UV spectra, on a Shimadzu UV-240; PMR spectra, on Bruker AM 200, 300, 400, and 500 FT NMR instruments; ¹³C NMR spectra, on a Bruker AM 500 FT NMR

at 50 and 125 MHz; mass spectra, on Varian-MAT 112S and Finnigan MAT 112 instruments; EI mass spectra, on a JEOL MAT 312 JMS HX-100 instrument. Optical rotations were determined on a JASCO DIP-360 instrument.

D-(+)-Galactose, $[M]^+$ 180, mp 158-160°C, $[\alpha]_D^{27}$ 83.6° (*c* 0.11, MeOH).

EI mass spectrum: 180 [M]⁺, 149 (2), 131 (3), 103 (13), 85 (11), 73 (100), 61 (68), 57 (48).

PMR spectrum (DMSO, 400 MHz, δ, ppm, J/Hz): 3.02 (1H, m), 3.09 (1H, m), 3.41 (2H, m), 3.53 (2H, m), 4.33 (1H, t, J = 5.7), 4.43 (1H, d, J = 6.5), 4.62 (1H, d, J = 4.2), 4.87 (1H, d, J = 5), 4.89 (1H, t, J = 3.7), 6.16 (1H, d, J = 4.2).

¹³C NMR spectrum (CD₃OD, 50 MHz): 61.275 (C-6), 70.657 (C-4), 71.955 (C-2), 72.411 (C-3), 73.151 (C-5), 92.264

(C-1).

Gallic acid, [M]⁺ 170, mp 221-223°C.

UV spectrum (MeOH, λ_{max} , nm): 270.

EI mass spectrum: 170 [M]⁺, 153 (86), 135 (11), 125 (23), 107 (8), 79 (25), 56 (11), 54 (30), 52 (50).

PMR spectrum (CD₃OD, 500 MHz, δ , ppm): 7.04 (2H, s, H-6 and H-8).

Syringic acid, [M]⁺ 198, mp 208-210°C.

UV spectrum (MeOH, λ_{max} , nm): 275.

PMR spectrum (CD₃OD, 500 MHz, δ, ppm): 7.01 (2H, s, H-6 and H-8), 3.82 (6H, s, OCH₃-3,5).

Ellagic acid, C₁₄H₆O₈, [M]⁺ 270, mp 360°C.

UV spectrum (MeOH, λ_{max} , nm): 257, 360.

The compound gives a red color with CH_3CO_2H (conc.) and $NaNO_2$, characteristic of free ellagic acid. The melting point of a mixed sample with ellagic acid was not depressed.

Quercetin, [M]⁺ 302, mp 310-312°C.

UV spectrum (MeOH, λ_{max} , nm): 256, 340 sh, 372.

PMR spectrum (CD₃OD, 200 MHz, δ, ppm, J/Hz): 6.17 (1H, d, J = 1.7, H-6), 6.38 (1H, d, J = 1.5, H-8), 6.86 (1H, d, J = 8.5, H-5'), 7.62 (1H, d, J = 8, H-2'), 7.7 (1H, d, J = 8, H-6').

Rutin (quercetin-3-O-[\alpha-L-rhamnopyranosyl-(1\rightarrow6)-\beta-D-glucopyranoside], [M]⁺ 609, mp 198-200°C. UV spectrum (MeOH, \lambda_{max}, nm): 262, 304, 374.

 $PMR \ spectrum \ (CD_3OD, \ 400 \ MHz, \ \delta, \ ppm, \ J/Hz): 0.84 \ (3H, \ d, \ J=6, \ CH_3-rhamn.), \ 3.29-3.80 \ (sugar \ region), \ 4.2 \ (1H, \ s, \ H-1''), \ 5.7 \ (1H, \ d, \ J=7.5, \ H-1'''), \ 6.08 \ (1H, \ d, \ J=2), \ 6.36 \ (1H, \ d, \ J=2), \ 6.78 \ (1H, \ d, \ J=9, \ H-5'), \ 7.62 \ (1H, \ d, \ J=2, \ H-2'), \ 5.7 \ (1H, \ d, \ J=7.5, \ H-1'''), \ 5.7 \ (1H, \ d, \ J=2), \ 6.78 \ (1H, \ d, \ J=9, \ H-5'), \ 7.62 \ (1H, \ d, \ J=2, \ H-2'), \ 7.62 \ (1H, \ d, \ J=2'), \ 7.62$

7.7 (1H, d, J = 2, H-6'). Acid hydrolysis produced myricetin, glucose, and rhamnose.

Myricetin, [M]⁺ 318, mp 364-366°C.

UV spectrum (MeOH, λ_{max} , nm): 252, 340 sh, 374. IR spectrum (KBr, λ_{max} , cm⁻¹): 3385-3300, 1660, 1565, 1516. PMR spectrum (CD₃OD, 500 MHz, δ , ppm, J/Hz): 6.17 (1H, d, J = 2), 6.37 (1H, d, J = 2), 7.34 (2H, s).

Myricetrin (myricetin 3-O-*α*-**L**-rhamnopyranoside), $[M]^+$ 364, mp 194-197°C, $[α]_D^{27}$ -122.6° (*c* 0.06, MeOH). UV spectrum (MeOH, $λ_{max}$, nm): 257, 352.

PMR spectrum (CD₃OD, 400 MHz, δ , ppm, J/Hz): 0.95 (3H, d, J = 6, CH₃-rhamn.), 3.25 (1H, t, H-5"), 3.62 (1H, dd, H-3"), 3.72 (1H, dd, H-2"), 4.98 (1H, dd, H-4"), 5.3 (1H, s, H-1"), 6.19 (1H, d, J = 2), 6.36 (1H, d, J = 2), 6.93 (2H, s).

¹³C NMR spectrum (DMSO, 125 MHz): 17.59 (CH₃-rhamn.), 70.718 (C-5"), 70.509 (C-2"), 70.647 (C-3"), 71.401 (C-4"), 93.692 (C-8), 98.844 (C-6), 102.019 (C-1"), 104.093 (C-10), 108.052 (C-2',6'), 119.73 (C-1'), 134.39 (C-3), 136.618 (C-4'), 145.859 (C-3',5'), 156.553 (C-9), 157.562 (C-2), 161.397 (C-5), 164.47 (C-7), 177.845 (C-4). Acid hydrolysis produced myricetin and rhamnose.

Myricetin 7-O- α -**L**-rhamnopyranoside, [M]⁺ 464, mp 178-180°C, $[\alpha]_D^{27}$ -135.3° (*c* 0.06, MeOH).

UV spectrum (MeOH, λ_{max}, nm): 262, 304 sh, 355; +NaOAc: 256, 389.

PMR spectrum (CD₃OD, 400 MHz, δ , ppm, J/Hz): 1.00 (3H, d, J = 6, CH₃), 3.29 (1H, m, H-5"), 3.54 (1H, m, H-3"), 3.88 (1H, dd, H-2"), 4.25 (1H, s, H-4"), 5.36 (1H, s, H-1"), 6.23 (1H, d, J = 2), 6.39 (1H, d, J = 2), 6.99 (2H, s). Acid hydrolysis produced myricetin and rhamnose.

Myricetin rutinoside (myricetin 3-O-[α -L-rhamnopyranosyl-(1 \rightarrow 6)- β -D-glucopyranoside], [M]⁺ 626, mp 190-192°C.

UV spectrum (MeOH, λ_{max} , nm): 256, 308, 363.

PMR spectrum (CD₃OD, 400 MHz, δ , ppm, J/Hz): 0.96 (3H, d, J = 6, CH₃-rhamn.), 3.12-4.56 (sugar region), 4.56 (1H, s, H-1"), 5.31 (1H, d, J = 6.5, H-1""), 6.19 (1H, d, J = 2), 6.35 (1H, d, J = 2), 6.94 (2H, s). Acid hydrolysis produced myricetin, glucose, and rhamnose.

Myricetin 3-O-*α*-**L**-arabinopyranoside, $[M]^+$ 450, mp 203-205°C, $[α]_D^{27}$ -112° (*c* 0.012, MeOH).

UV spectrum (MeOH, λ_{max} , nm): 257, 363.

PMR spectrum (CD₃OD, 500 MHz, δ , ppm, J/Hz): 3.45 (1H, d, J = 11.6, H-4"), 3.63 (1H, dd, J = 3, H-5"), 3.82-3.90 (2H, m, H-2" and H-3"), 5.17 (1H, d, J = 6.7, H-1"), 6.19 (1H, d, J = 2, H-6), 6.35 (1H, d, J = 2, H-8), 7.3 (2H, s, H-2' and H-6'). Acid hydrolysis produced myricetin and arabinose.

Myricetin 3-O-β-D-xylopyranoside, $[M]^+$ 450, mp 185-187°C, $[\alpha]_D^{27}$ -28° (*c* 0.04, MeOH).

UV spectrum (MeOH, λ_{max} , nm): 256, 284 sh, 360.

PMR spectrum (CD₃OD, 200 MHz, δ , ppm, J/Hz): 3.51 (1H, dd, J = 3, H-5"), 3.68 (1H, dd, J = 3, H-4"), 3.82-3.90 (2H, m, H-2" and H-3"), 5.21 (1H, d, J = 5.8, H-1"), 6.23 (1H, d, J = 2), 6.42 (1H, d, J = 2), 7.34 (2H, s). Acid hydrolysis produced myricetin and xylose.

Myricetin 3-O- α -**L**-(2"-O- α -rhamnopyranosyl)rhamnopyranoside, [M]⁺ 611, mp 183-185°C.

UV spectrum (MeOH, λ_{max} , nm): 258, 302 sh, 354; +AlCl₃: 272, 309, 424; +AlCl₃/HCl: 273, 307, 404; +NaOAc: 262, 299, 374.

PMR spectrum (CD₃OD, 400 MHz, δ, ppm, J/Hz): 0.97 (3H, d, J = 6, H-6^{''}), 1.27 (3H, d, J = 6, H-6^{''}), 3.15 (1H, m, H-4^{'''}), 3.32 (1H, m, H-4^{'''}), 3.50-3.57 (2H, m, H-5^{'''}), 3.75-3.79 (2H, m, H-3^{'''}), 4.20-4.24 (2H, t, H-2^{'''}), 4.55 (1H, br.s, H-1^{'''}), 5.31 (1H, br.s, H-1^{''}), 6.19 (1H, d, J = 1.8, H-6), 6.36 (1H, d, J = 1.8, H-8), 6.94 (2H, s, H-2', H-6'). Acid hydrolysis produced myricetin and rhamnose.

Myricetin 3-O-\beta-D-(2'-galloyl)arabinopyranoside, C₂₇H₂₂O₁₆, [M]⁺ 602, mp 182-184°C, $[\alpha]_D^{27}$ -36° (*c* 0.05, MeOH).

UV spectrum (MeOH, λ_{max}, nm): 265, 366; +AlCl₃: 272, 431; +AlCl₃/HCl: 269, 411; +NaOAc: 271, 388.

PMR spectrum (CD₃OD, 500 MHz, δ, ppm, J/Hz): 3.46 (1H, d, J = 9, H-4''), 3.63 (1H, dd, J = 2.5, H-5''), 3.82-3.91 (2H, m, H-2'', H-3''), 5.17 (1H, d, J = 6.6, H-1''), 6.19 (1H, d, J = 1.8, H-6), 6.38 (1H, d, J = 1.8, H-8), 7.08 (2H, s, gallic acid H-2''', H-6''), 7.32 (2H, s, H-2', H-6').

¹³C NMR spectrum (DMSO, 50 MHz): 64.57 (C-5″), 66.313 (C-3″), 70.78 (C-4″), 71.896 (C-2″), 93.57 (C-8), 98.814 (C-6), 101.824 (C-1″), 103.99 (C-10), 108.59 (C-2′, C-6′), 108.876 (C-2″', C-6″), 119.97 (C-1′), 120.71 (C-1″), 134.021 (C-3), 136.91 (C-4′), 138.07 (C-4″), 145.64 (C-3″', C-5″), 145.483 (C-3″', C-5′), 156.387 (C-2), 156.539 (C-9), 161.304 (C-5), 164.35 (C-7), 167.67 (C=O), 177.605 (C-4). Acid hydrolysis produced myricetin, gallic acid, and arabinose.

(-)-Epigallocatechin-3-O-gallate, C₂₂H₁₈O₁₁, [M]⁺ 458, mp 140-142°C.

UV spectrum (MeOH, λ_{max} , nm): 274; +AlCl₃: 295; +NaOAc: 281.

IR spectrum (KBr, λ_{max} , cm⁻¹): 3419, 2526, 1692, 1615, 1534, 1448.

PMR spectrum (CD₃OD, 500 MHz, δ, ppm, J/Hz): 2.79-2.86 (1H, dd, J = 2.3, H-4α), 2.94-3.02 (1H, dd, J = 4.5, H-4β), 4.96 (1H, s, H-2), 5.52 (1H, s, H-3), 5.56 (1H, d, J = 1.5, H-6), 5.59 (1H, d, J = 1.5, H-8), 6.49 (2H, s, H-2', H-6'), 6.94 (2H, s, H-2'', H-6'').

¹³C NMR spectrum (DMSO, 125 MHz): 26.715 (C-4), 69.87 (C-3), 78.461 (C-2), 95.844 (C-8), 96.492 (C-6), 99.38 (C-10), 106.835 (C-2', C-6'), 110.213 (C-2", C-6"), 121.39 (C-1"), 130.733 (C-1'), 133.65 (C-4"), 146.14 (C-3", C-5"), 146.53 (C-3', C-5'), 157.64 (C-5), 157.67 (C-7, C-9), 167.603 (C=O).

3,5,7,3',4',6'-Hexahydroxyflavone, [M]⁺ 318, mp 306-308°C.

UV spectrum (MeOH, λ_{max} , nm): 256, 301 sh, 377; +AlCl₃: 271, 460; +AlCl₃/HCl: 266, 355 sh, 433; +NaOAc: 270, 390; +H₃BO₃: 261, 387.

IR spectrum (KBr, λ_{max} , cm⁻¹): 3385-3300, 1660, 1565, 1516.

Mass spectrum (EI, *m/z*): 318 (100) [M]⁺, 301 (17), 269 (4), 244 (8), 152 (38), 135 (18), 107 (10), 64 (49).

PMR spectrum (CD₃OD, 400 MHz, δ , ppm, J/Hz): 6.16 (1H, d, J = 1.6, H-6), 6.41 (1H, d, J = 2, H-8), 7.66 (1H, d, J = 1.6, H-5'), 7.75 (1H, d, J = 1.6, H-2').

¹³C NMR spectrum (DMSO, 125 MHz): 93.35 (d, C-8), 98.25 (d, C-6), 102.98 (C-10), 111.54 (d, C-2'), 114.092 (d, C-5'), 121.06 (C-1'), 136.12 (C-3), 140.11 (C-4'), 141.18 (C-3'), 146.13 (C-6'), 146.55 (C-2), 156.113 (C-9), 160.69 (C-5), 164.02 (C-7), 175.86 (C-4).

6"-Galloylmyricetin 3-O-β-D-galactopyranoside, $C_{28}H_{24}O_{17}$, [M]⁺632, mp 212-216°C, [α]_D²⁷ 16° (*c* 0.07, MeOH). UV spectrum (MeOH, λ_{max} , nm): 266, 366; +AlCl₃: 274, 435; +AlCl₃/HCl: 274, 412.

PMR spectrum (CD₃OD, 400 MHz, δ, ppm, J/H): 3.59 (1H, dd, J = 3.3, H-5"), 3.78-3.9 (3H, m, H-3", H-6"a, H-2"), 4.23-4.31 (2H, m, H-6"b, H-4"), 5.16 (1H, d, J = 7.8, H-1"), 6.15 (1H, d, J = 1.8, H-6), 6.34 (1H, d, J = 1.8, H-8), 6.89 (2H, s, gallic acid H-2", H-6"), 7.34 (2H, s, H-2', H-6').

¹³C NMR spectrum (DMSO, 125 MHz): 62.107 (C-6"), 68.043 (C-2"), 71.181 (C-3"), 72.642 (C-4"), 73.113 (C-5"), 94.013 (C-8), 99.597 (C-6), 102.627 (C-1"), 103.018 (C-10), 108.67 (C-2', C-6'), 108.778 (C-2"', C-6"'), 118.851 (C-1'), 119.699 (C-1"'), 133.611 (C-3), 137.520 (C-4'), 139.245 (C-4"'), 145.714 (C-3', C-5'), 145.803 (C-3"', C-5"'), 156.13 (C-2), 156.558 (C-9), 161.14 (C-5), 165.707 (C-7), 166.91 (C=O), 176.99 (C-4). Acid hydrolysis produced myricetin, gallic acid, and galactose.

Myricetin 3-O-β-D-galactopyranoside, [M]+ 480, mp 196-198°C.

UV spectrum (MeOH, λ_{max}, nm): 256, 308, 362; +AlCl₃: 274, 310, 430.

PMR spectrum (CD₃OD, 300 MHz, δ, ppm, J/Hz): 3.59 (1H, dd, J = 3.3, H-5"), 3.78-3.9 (3H, m, H-3", H-6"α, H-2"), 4.23-4.31 (2H, m, H-6"β, H-4"), 5.25 (1H, d, J = 7.9, H-1"), 6.17 (1H, d, J = 1.8, H-6), 6.42 (1H, d, J = 1.8, H-8), 7.23 (2H, s, H-2', H-6'). Acid hydrolysis produced myricetin and galactose.

Myricetin 3-O- β **-D-glucopyranoside**, [M]⁺ 480, mp 275-277°C.

UV spectrum (MeOH, λ_{max}, nm): 256, 308, 364; +AlCl₃: 272, 310, 430.

PMR spectrum (CD₃OD, 300 MHz, δ , ppm, J/Hz): 3.31-3.51 (3H, m), 3.58 (1H, dd), 3.64 (1H, dd), 5.27 (1H, d, J = 7.9, H-1"), 6.20 (1H, d, J = 1.8, H-6), 6.39 (1H, d, J = 1.8, H-8), 7.29 (2H, s, H-2', H-6'). Acid hydrolysis produced myricetin and glucose.

6"-Galloylmyricetin 3-O-β-D-glucopyranoside, $C_{28}H_{24}O_{17}$, [M]⁺ 632, [α]_D²⁷ -51° (*c* 0.08, MeOH).

UV spectrum (MeOH, λ_{max}, nm): 266, 301, 362; +AlCl₃: 276, 440; +AlCl₃/HCl: 274, 412.

PMR spectrum (CD₃OD, 400 MHz, δ , ppm, J/Hz): 3.42 (2H, m), 3.55 (1H, t, J = 8), 4.25 (1H, dd, J = 1.5), 4.37 (1H, dd, J = 5.7), 5.16 (1H, d, J = 7.8, H-1"), 6.15 (1H, d, J = 1.8, H-6), 6.34 (1H, d, J = 1.8, H-8), 6.89 (2H, s, gallic acid H-2", H-6"), 7.34 (2H, s, H-2', H-6').

¹³C NMR spectrum (DMSO, 125 MHz): 63.798 (C-6"), 69.937 (C-4"), 74.008 (C-5"), 74.578 (C-2"), 76.521 (C-3"), 93.419 (C-8), 98.676 (C-6), 101.256 (C-1"), 103.901 (C-10), 108.593 (C-2', C-6'), 108.697 (C-2", C-6"), 119.393 (C-1'), 120.069 (C-1"), 133.635 (C-3), 136.714 (C-4'), 138.393 (C-4"), 145.371 (C-3', C-5', C-3"', C-5"'), 156.050 (C-2), 156.216 (C-9), 161.186 (C-5), 164.060 (C-7), 165.777 (C=O), 177.263 (C-4). Acid hydrolysis produced myricetin, gallic acid, and glucose.

REFERENCES

- 1. P. W. Agrawal, *Phytochemistry*, **31**, 3307 (1992).
- 2. I. G. Abdulladzhanova, S. M. Mavlyanov, and D. N. Dalimov, *Khim. Prir. Soedin.*, 167 (2001).
- 3. S. S. N. Murthy, N. S. Prakkasa Rao, and A. S. R. Anganeyulu, *Curr. Sci.*, 43, 341 (1974).
- 4. E. C. Bate-Smith, *Phytochemistry*, **16**, 1421 (1977).
- 5. M. S. Luk'yanchikov and A. L. Kazakov, *Khim. Prir. Soedin.*, 251 (1982).
- 6. K. B. Kobakhidze and M. D. Alaniya, *Khim. Prir. Soedin.*, 162 (2002).
- 7. W. H. Parker and B. A. Bohm, *Phytochemistry*, **14**, 553 (1975).
- 8. C. C. Shen, Y. S. Chang, and L. K. Ho, *Phytochemistry*, **34**, 843 (1993).
- 9. K. R. Markham, B. Ternai, R. Stanley, H. Geiger, and T. J. Mabry, *Tetrahedron*, **34**, 1389 (1978).
- 10. G. G. Zapesochnaya, Khim. Prir. Soedin., 695 (1982).
- 11. L. O. M. Arot, J. O. Midiwo, and W. Kraust, *Phytochemistry*, **43**, 1107 (1996).
- 12. S. Kadota, Y. Takamori, K. N. Nyein, T. Kikuchi, K. Tanaka, and H. Ekimoto, *Chem. Pharm. Bull.*, **38**, 2687 (1990).
- 13. J. M. Miller and B. A. Bohm, *Phytochemistry*, **18**, 1412 (1979).
- 14. A. Braca, A. R. Bilia, J. Mendez, and I. Morelli, *Phytochemistry*, **51**, 1125 (1999).
- 15. A. L. Davis, Y. Cai, A. P. Davies, and J. R. Lewis, Magn. Reson. Chem., 34, 887 (1996).
- 16. J. Kagan, *Phytochemistry*, **6**, 317 (1967).
- 17. I. Merfort and D. Wendisch, *Planta Med.*, 434 (1987).

- 18. H. H. Barakat, A. M. Souleman, S. A. Hussein, O. A. Ibrahiem, and M. A. Nawwar, *Phytochemistry*, **51**, 139 (1999).
- 19. J. B. Harborne, ed., Methods in Plant Biochemistry, Vol. 1: Plant Phenolics, Academic, London (1989), pp. 2-27.
- 20. T. J. Mabry and K. R. Markham, in: *Flavonoids*, J. B. Harborne, T. J. Mabry, and H. Mabry, eds., Academic, New York (1975), Vol. 1, pp. 78-126.
- 21. K. R. Markham, *Techniques of Flavonoids Identification*, Academic, London (1982), pp. 45-77.
- 22. K. A. Pawan, S. T. Ragnunath, and C. B. Manesh, in: *Studies in Organic Chemistry. 39. Carbon-13 NMR of Flavonoids*, P. K. Agrawal, ed., Elsevier, Amsterdam (1989), pp. 97-158.